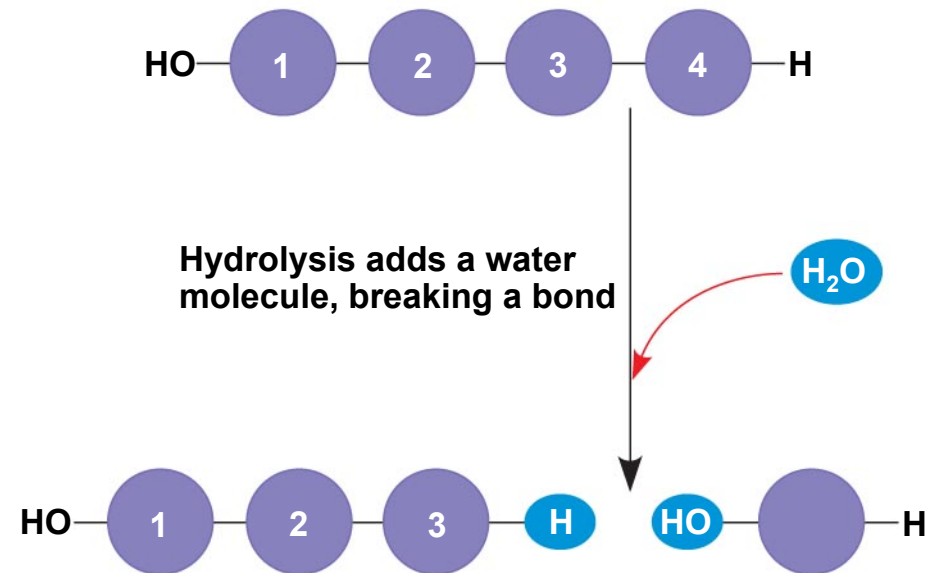
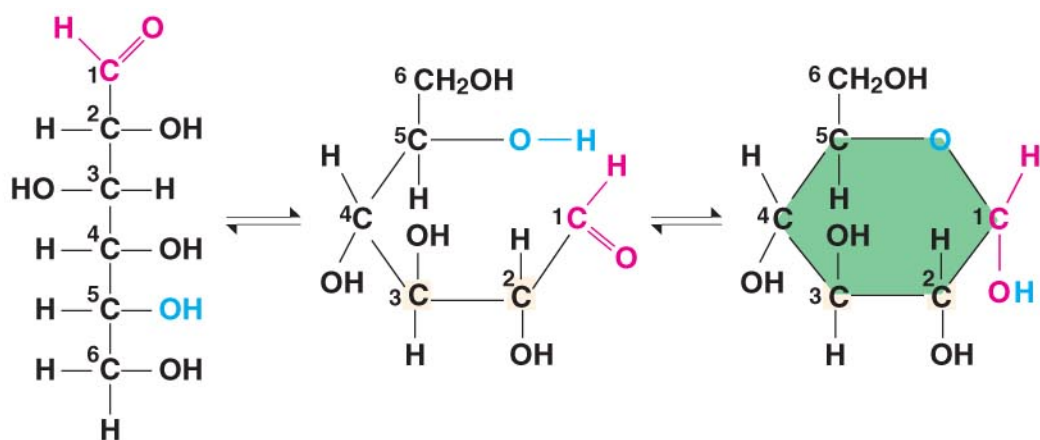


(a) Dehydration reaction in the synthesis of a polymer



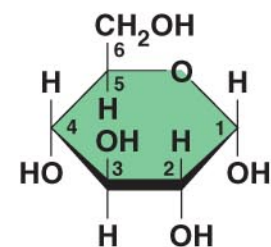
(b) Hydrolysis of a polymer

	Trioses ($C_3H_6O_3$)	Pentoses ($C_5H_{10}O_5$)	Hexoses ($C_6H_{12}O_6$)	
Aldoses	$ \begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Glyceraldehyde</p>	$ \begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Ribose</p>	$ \begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Glucose</p>	$ \begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Galactose</p>
Ketoses	$ \begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Dihydroxyacetone</p>	$ \begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Ribulose</p>	$ \begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H} \end{array} $ <p>Fructose</p>	

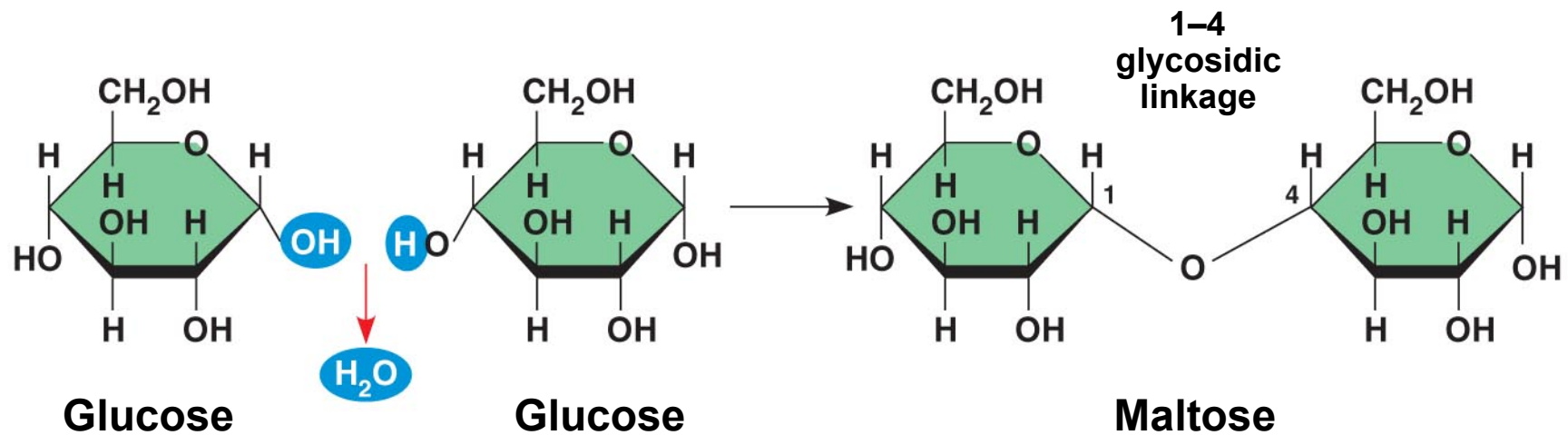


(a) Linear and ring forms

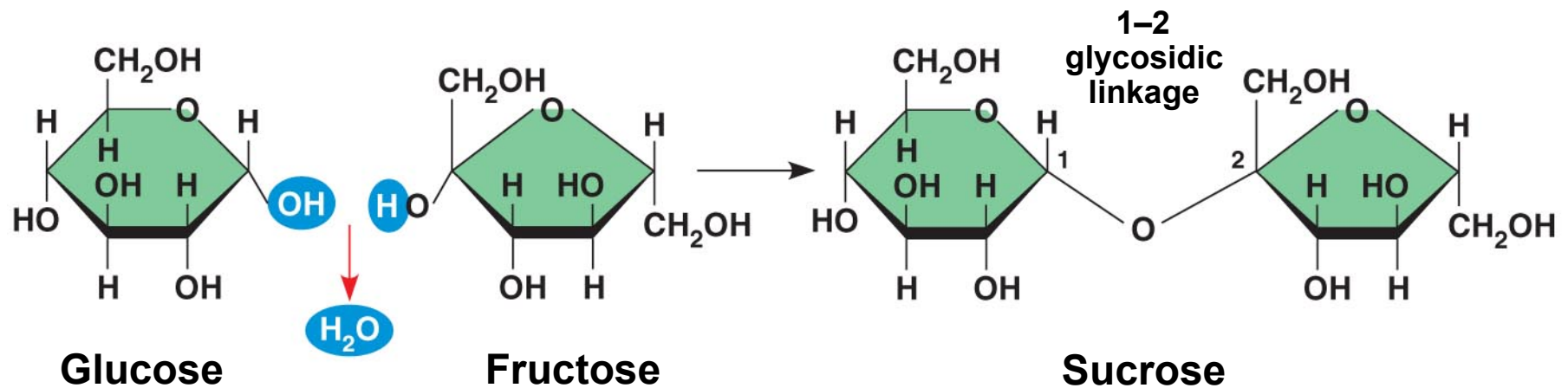
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(b) Abbreviated ring structure

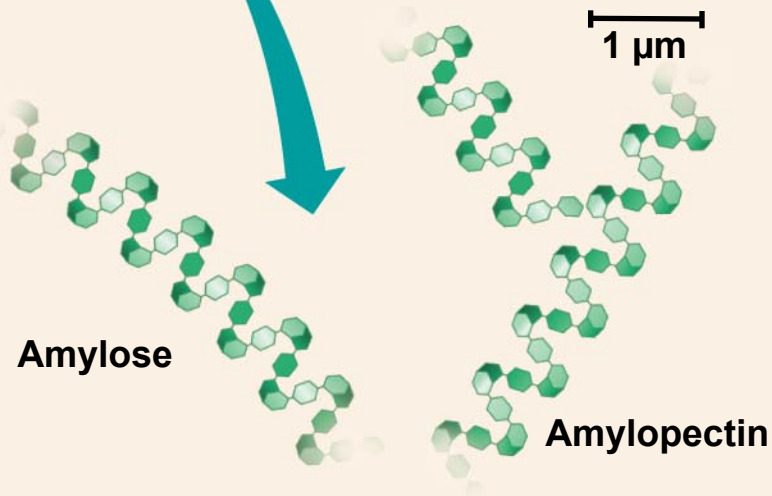
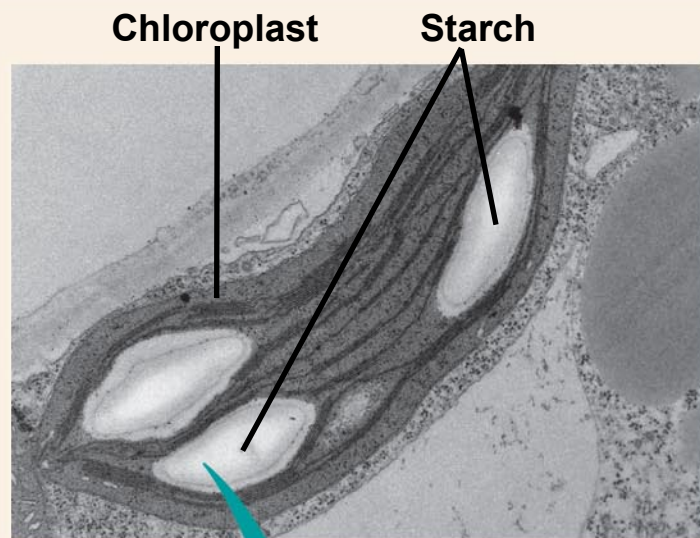


(a) Dehydration reaction in the synthesis of maltose

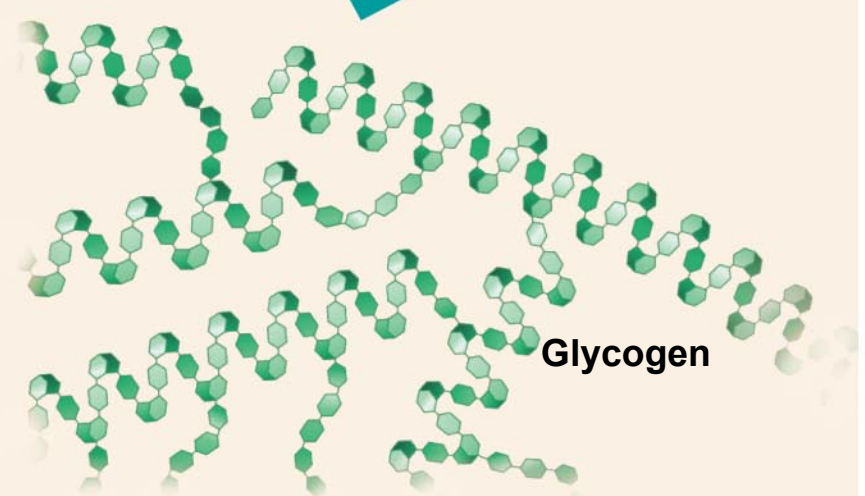
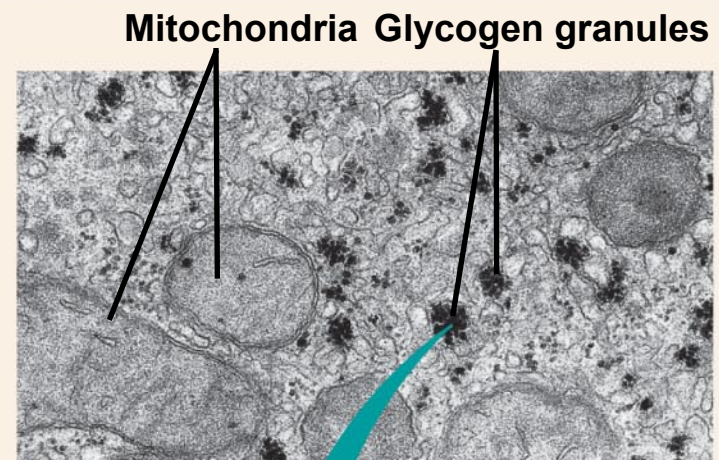


(b) Dehydration reaction in the synthesis of sucrose

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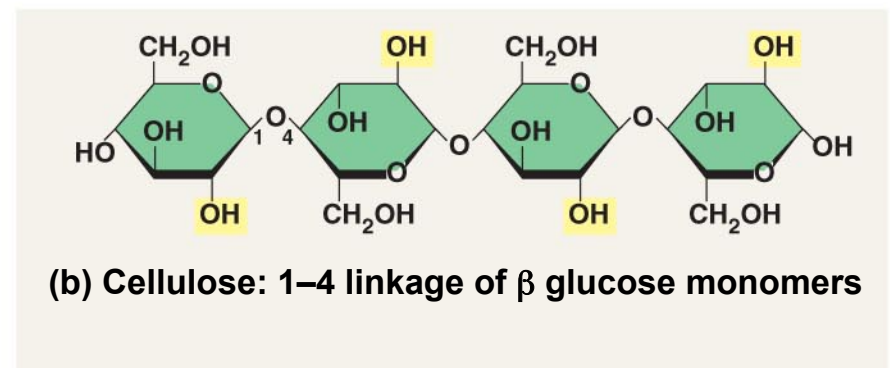
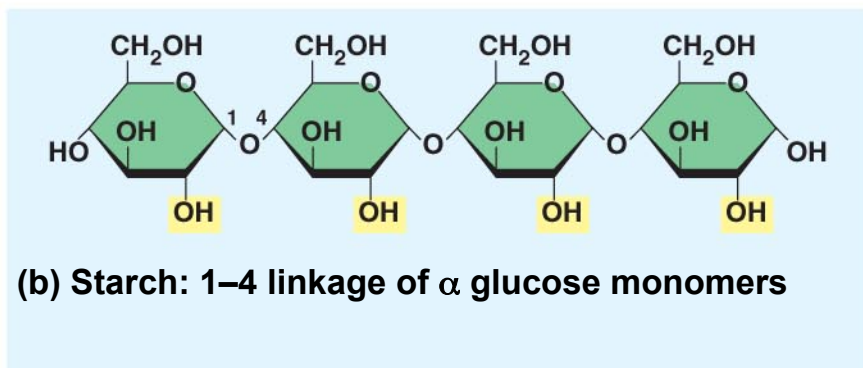
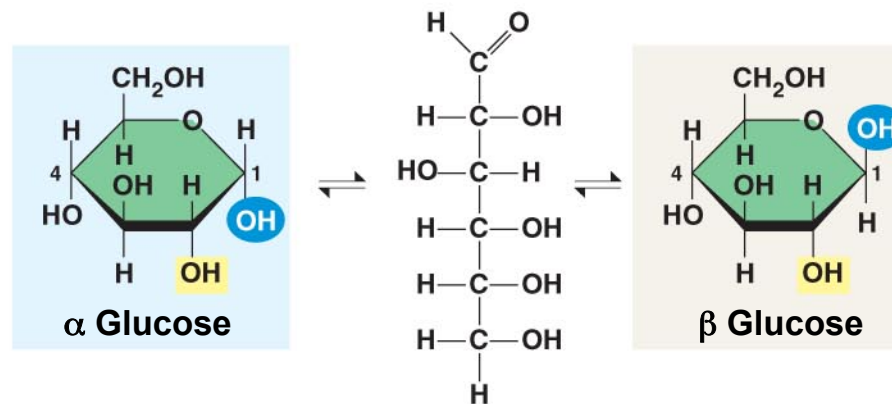


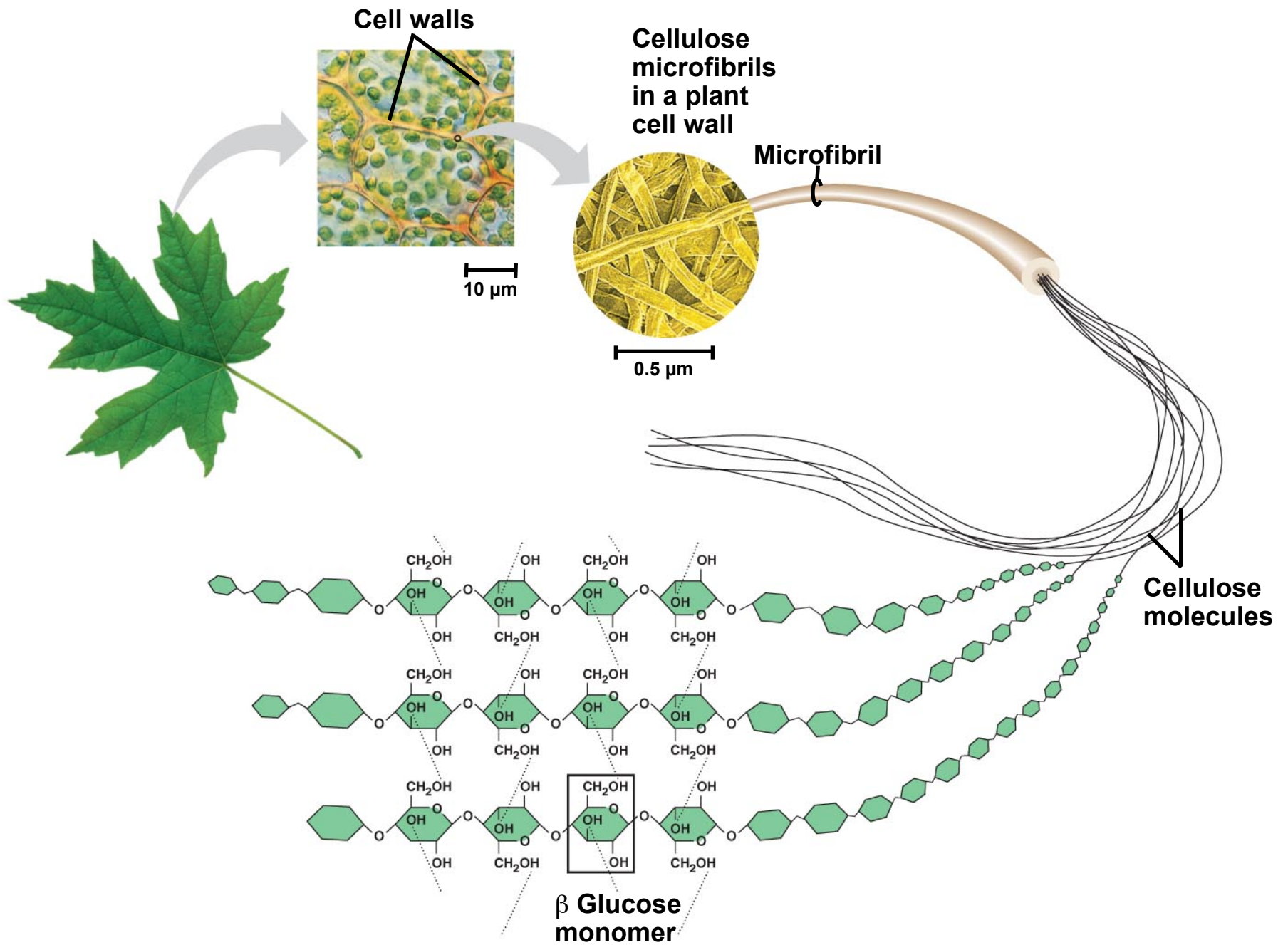
(a) Starch: a plant polysaccharide



(b) Glycogen: an animal polysaccharide

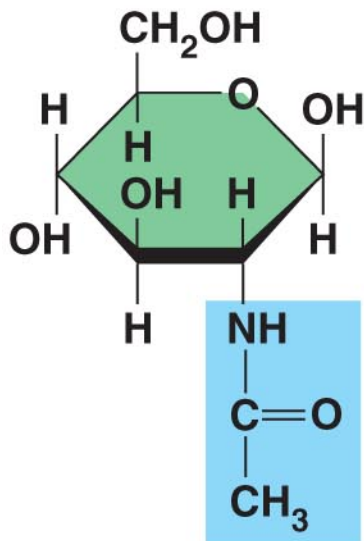
(a) α and β glucose ring structures







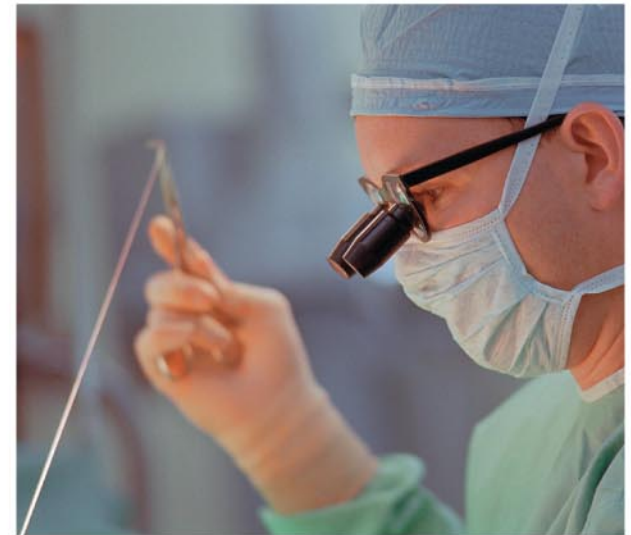
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(a) The structure of the chitin monomer.

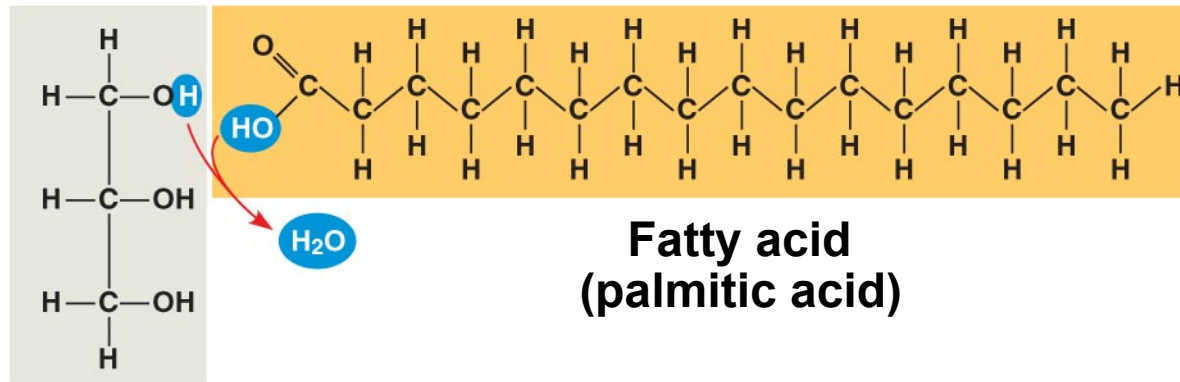


(b) Chitin forms the exoskeleton of arthropods.



(c) Chitin is used to make a strong and flexible surgical thread.

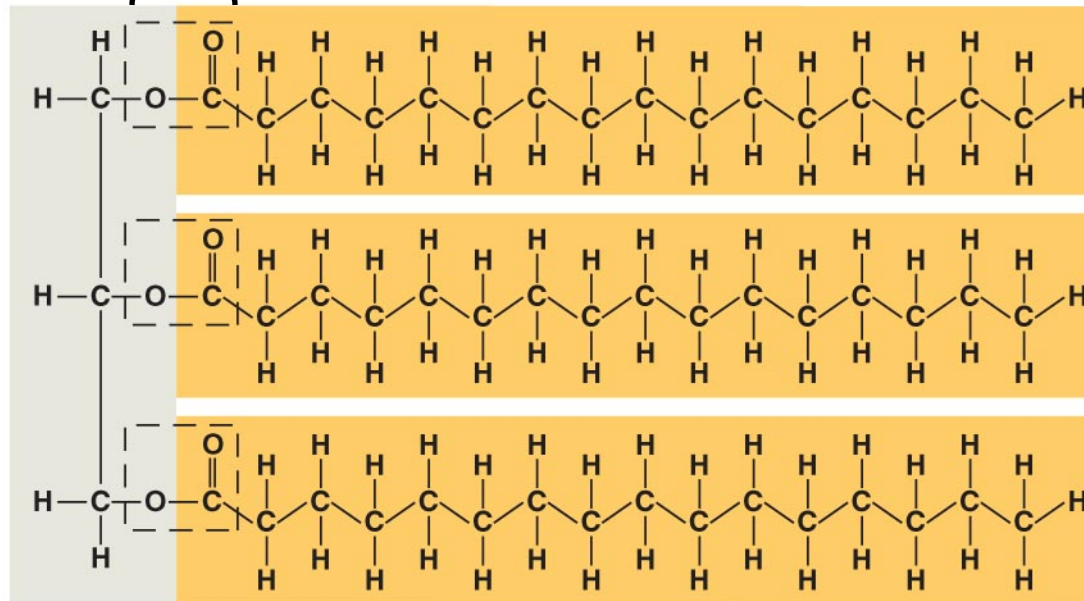
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Glycerol

(a) Dehydration reaction in the synthesis of a fat

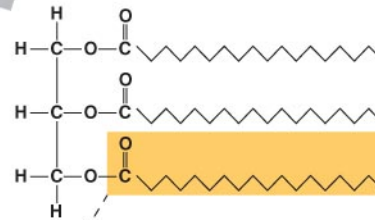
Ester linkage



(b) Fat molecule (triacylglycerol)

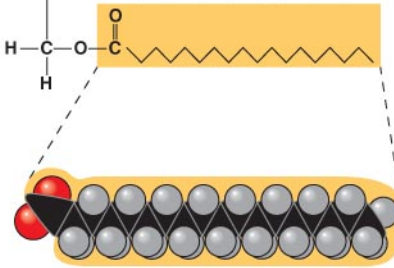


Structural formula of a saturated fat molecule

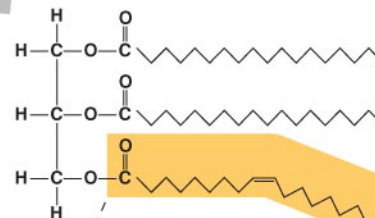


Stearic acid, a saturated fatty acid

(a) Saturated fat

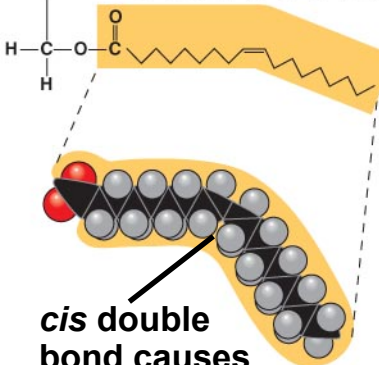


Structural formula of an unsaturated fat molecule

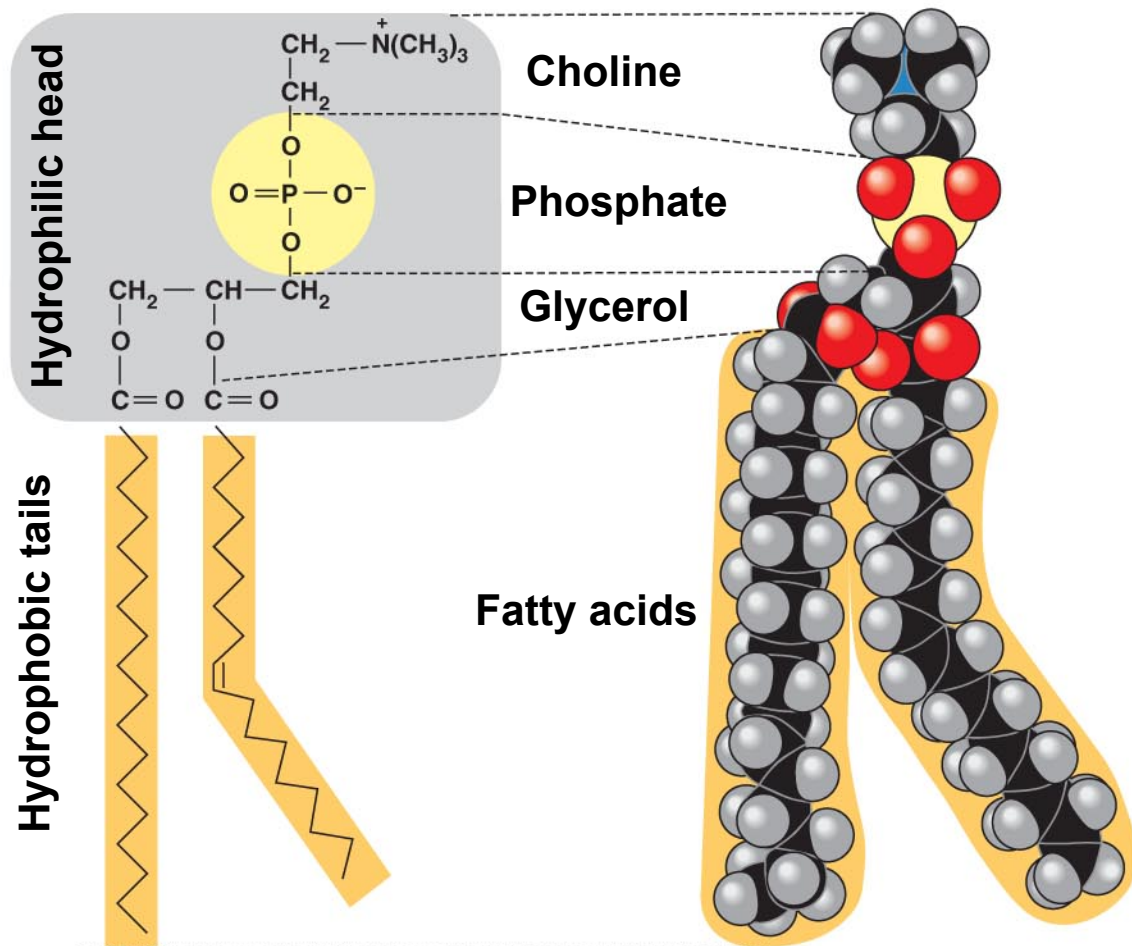


Oleic acid, an unsaturated fatty acid

(b) Unsaturated fat

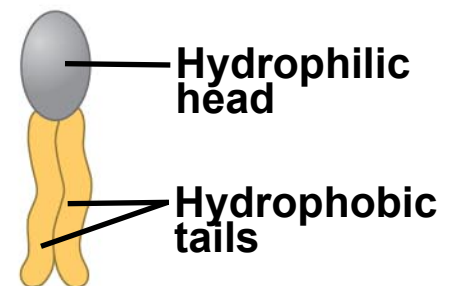


cis double bond causes bending



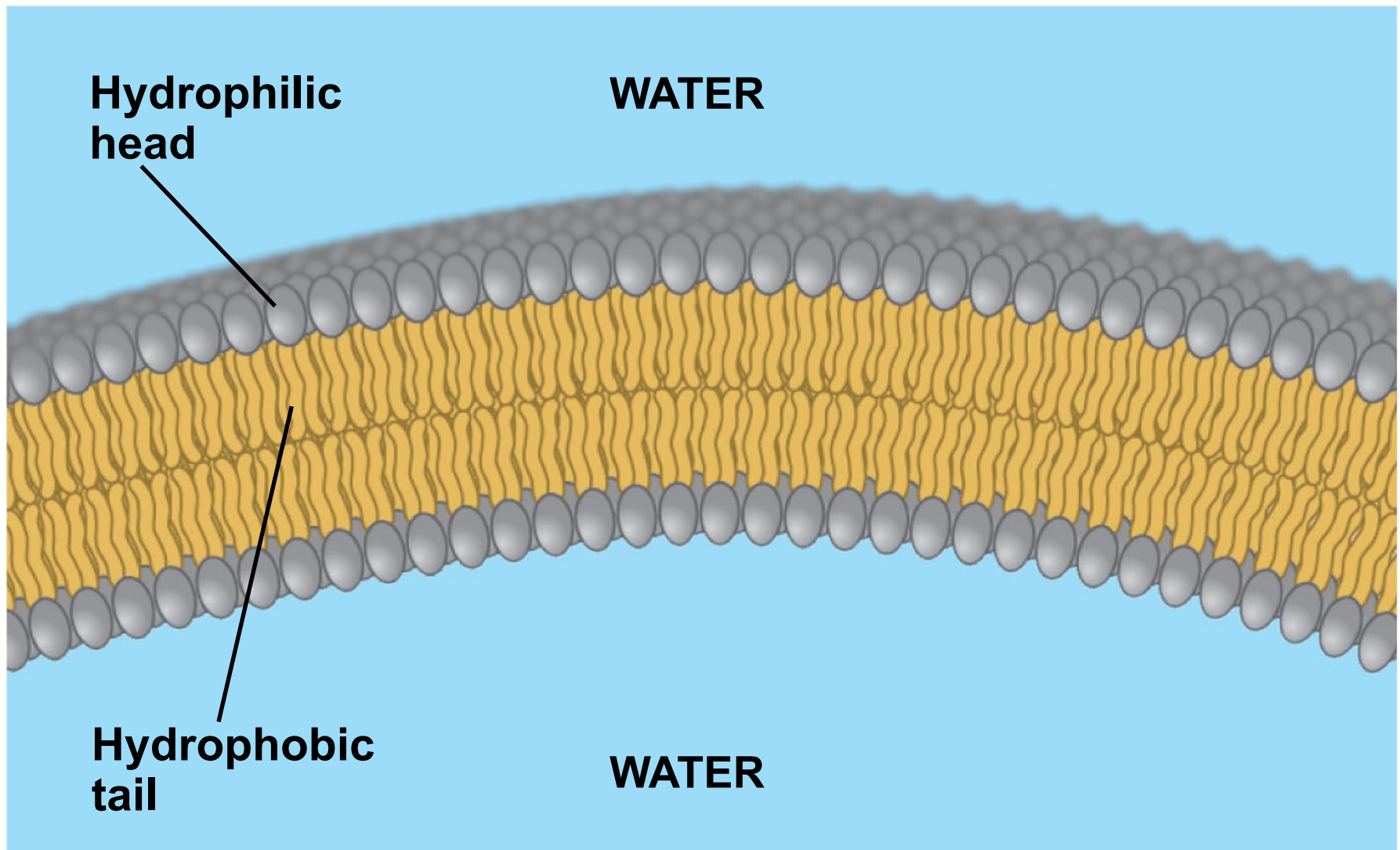
(a) Structural formula

(b) Space-filling model

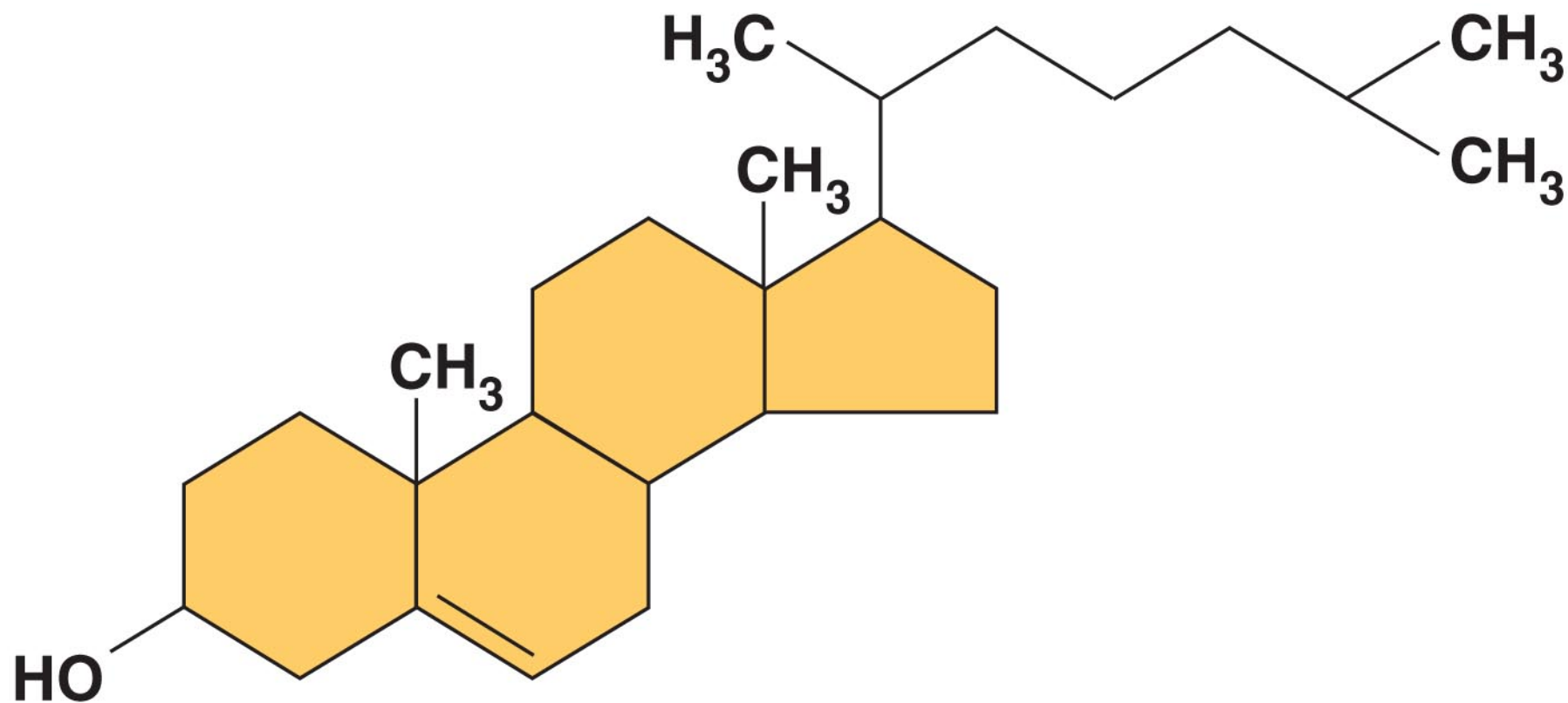


(c) Phospholipid symbol

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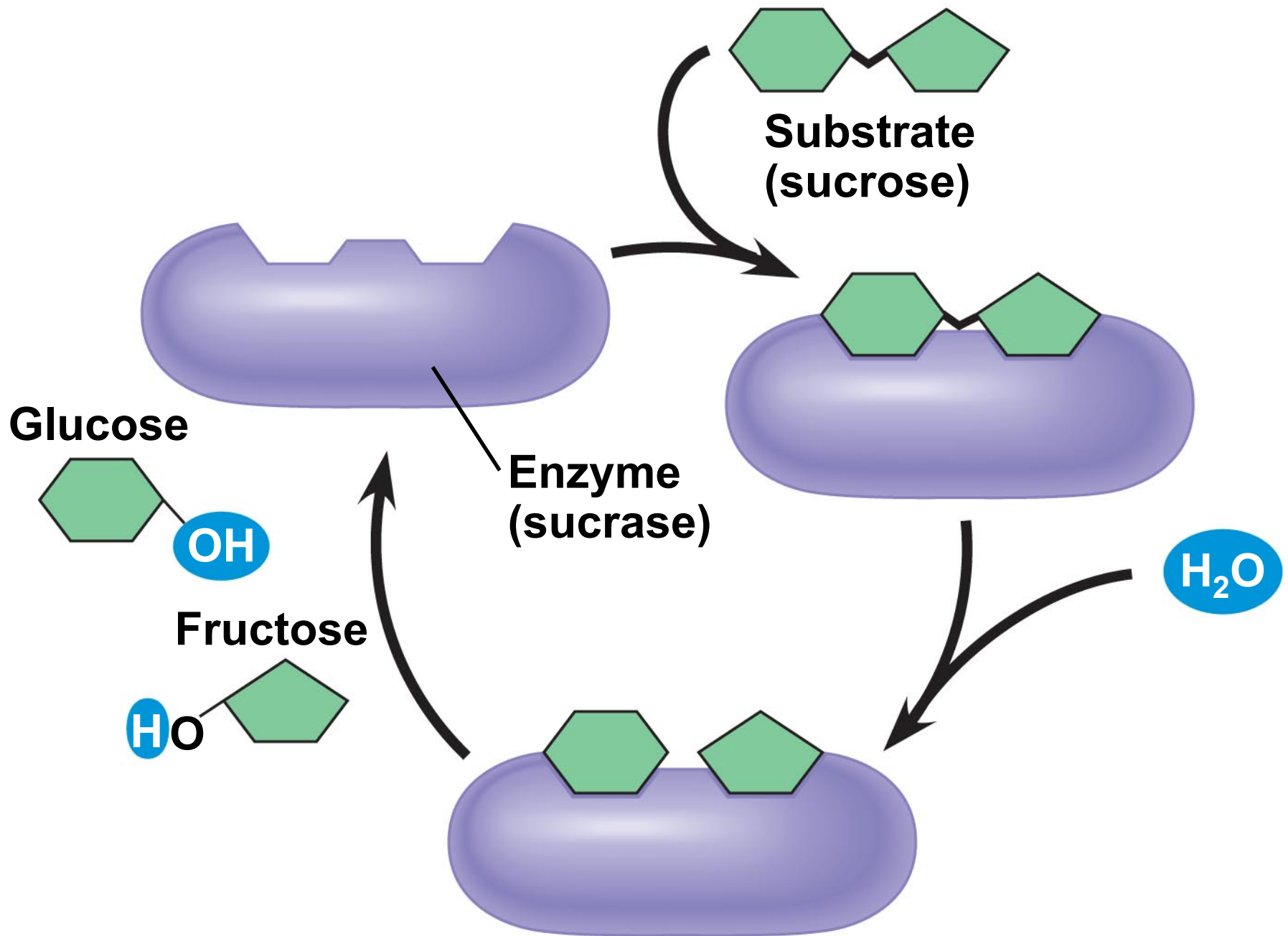


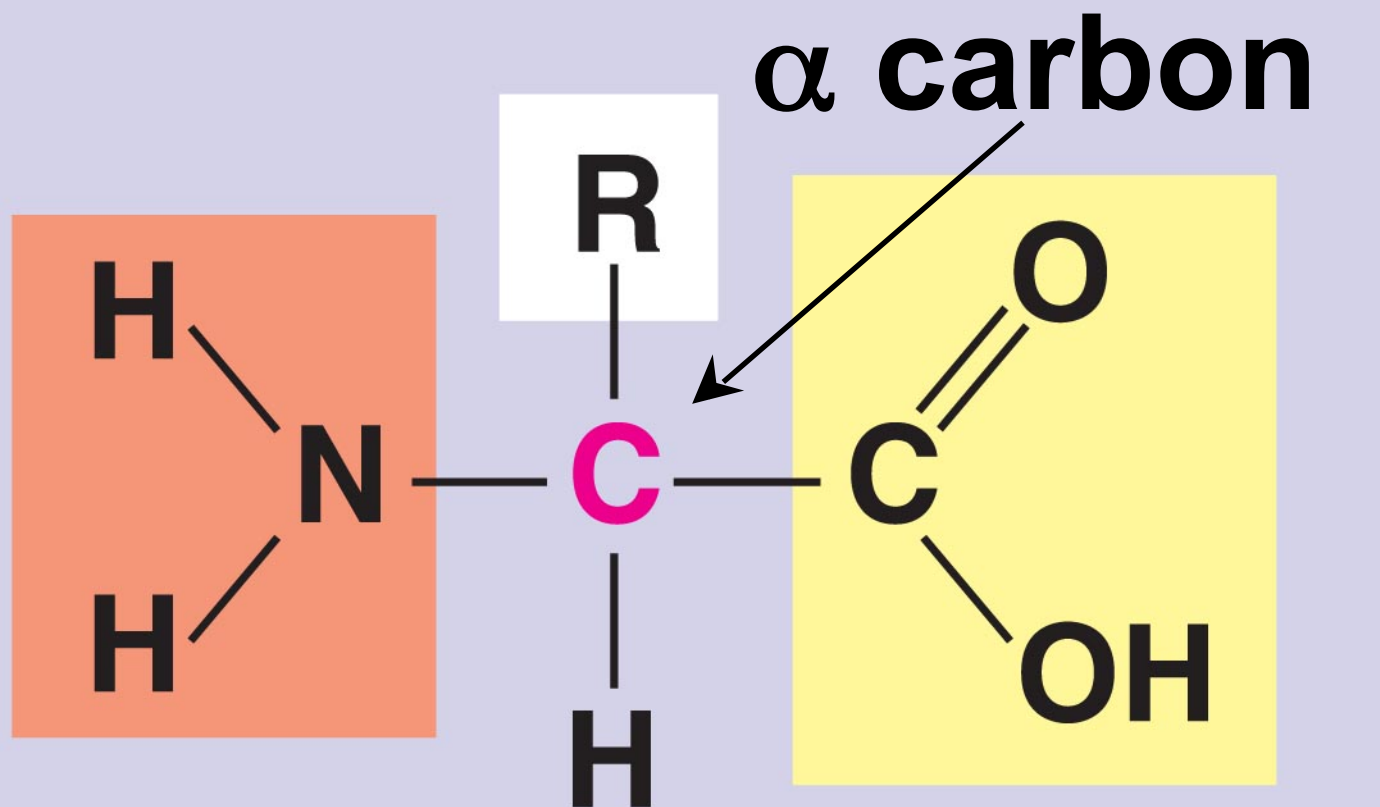
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Table 5.1 An Overview of Protein Functions

Type of Protein	Function	Examples
Enzymatic proteins	Selective acceleration of chemical reactions	Digestive enzymes
Structural proteins	Support	Silk fibers; collagen and elastin in animal connective tissues; keratin in hair, horns, feathers, and other skin appendages
Storage proteins	Storage of amino acids	Ovalbumin in egg white; casein, the protein of milk; storage proteins in plant seeds
Transport proteins	Transport of other substances	Hemoglobin, transport proteins
Hormonal proteins	Coordination of an organism's activities	Insulin, a hormone secreted by the pancreas
Receptor proteins	Response of cell to chemical stimuli	Receptors in nerve cell membranes
Contractile and motor proteins	Movement	Actin and myosin in muscles, proteins in cilia and flagella
Defensive proteins	Protection against disease	Antibodies combat bacteria and viruses.

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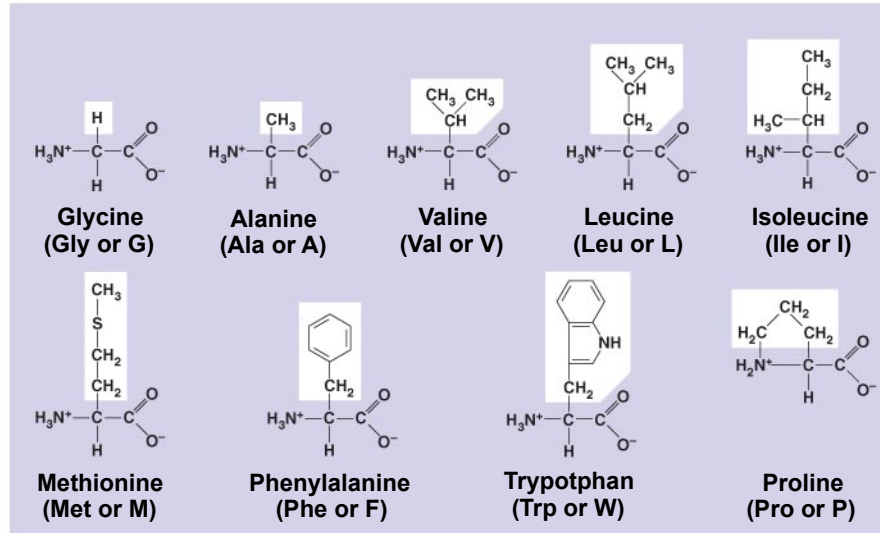




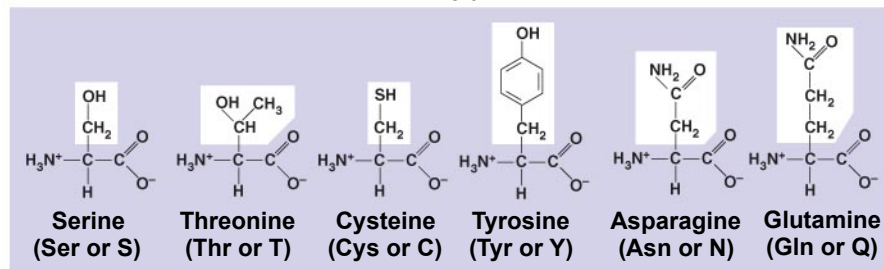
**Amino
group**

**Carboxyl
group**

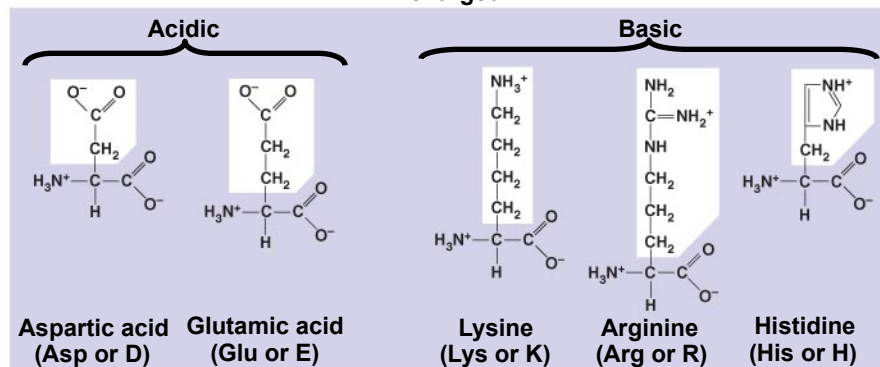
Nonpolar

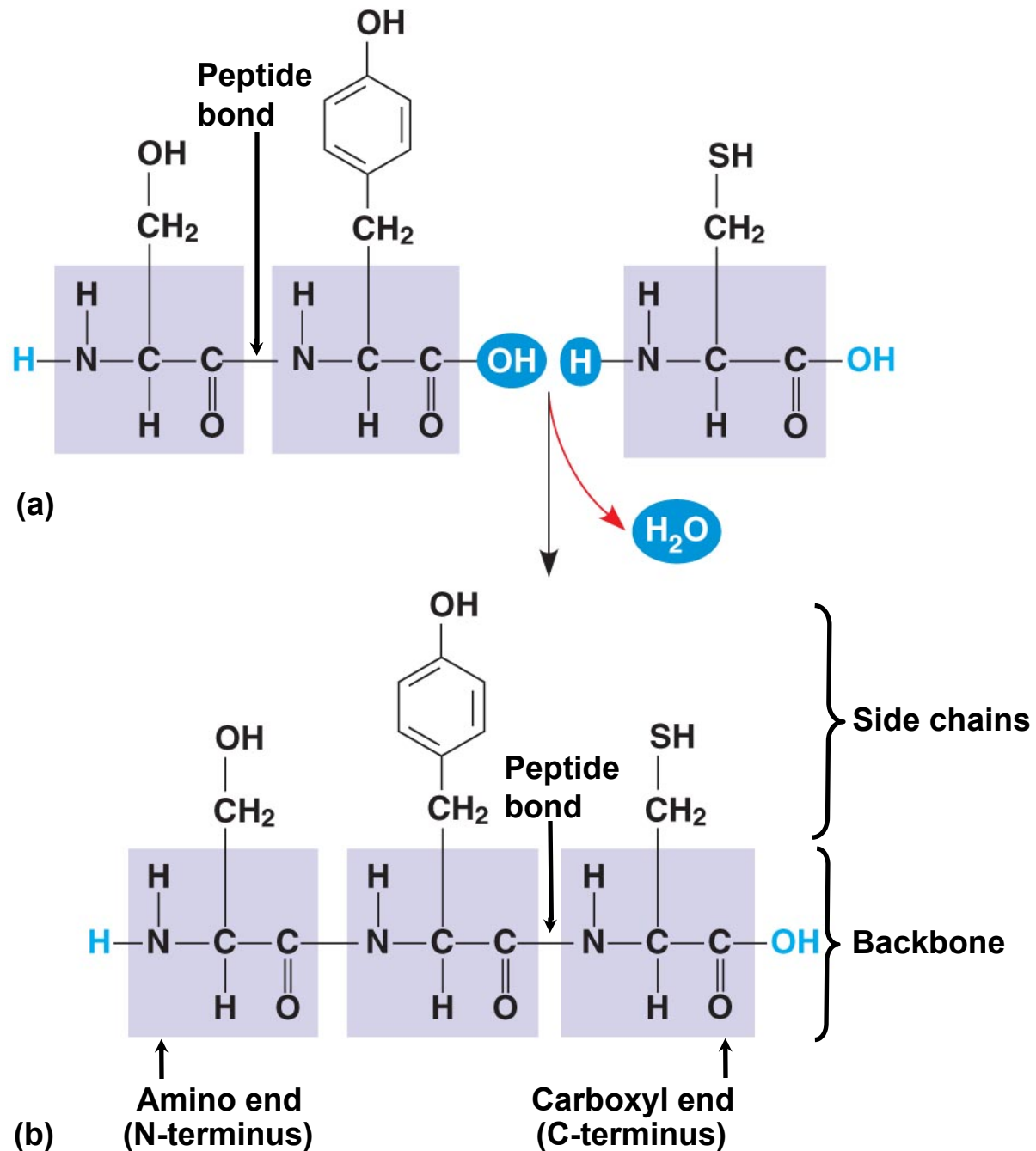


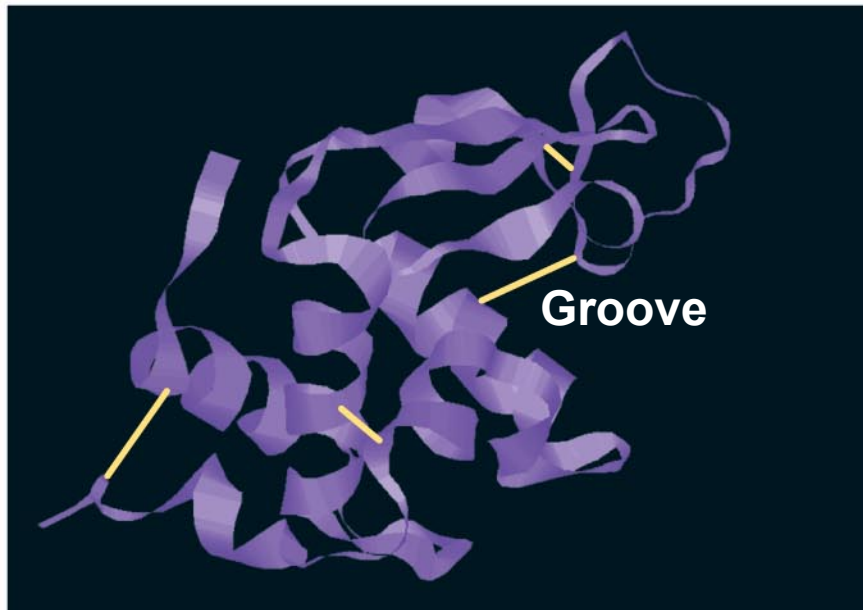
Polar



Electrically charged

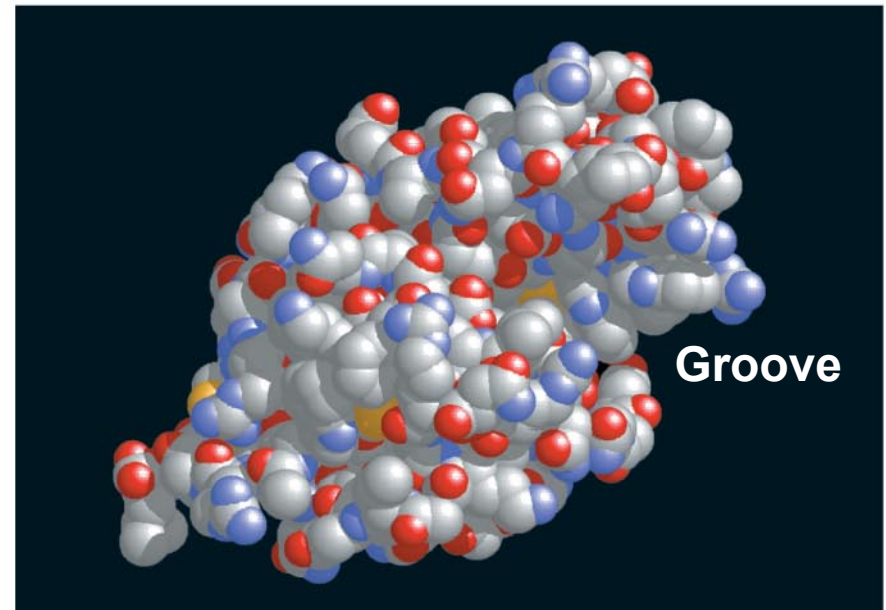






(a) A ribbon model of lysozyme

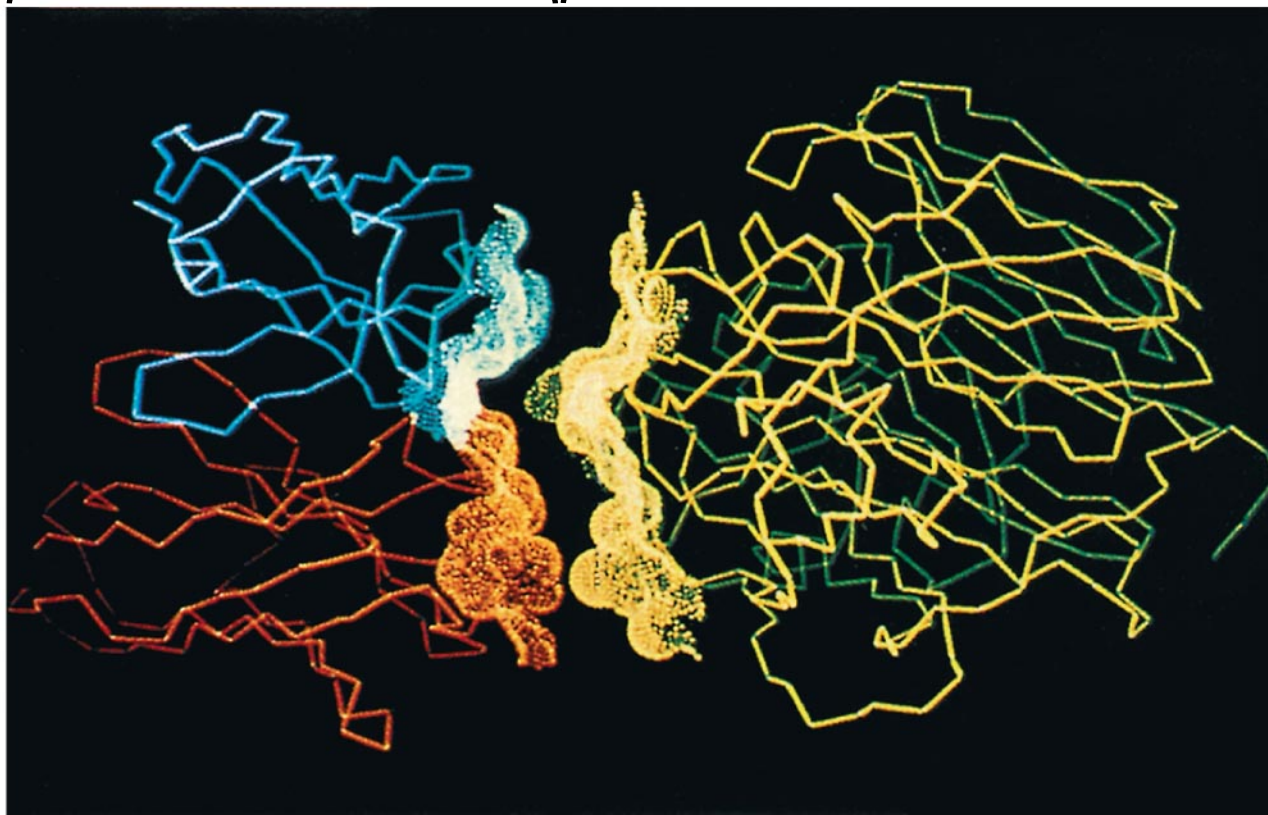
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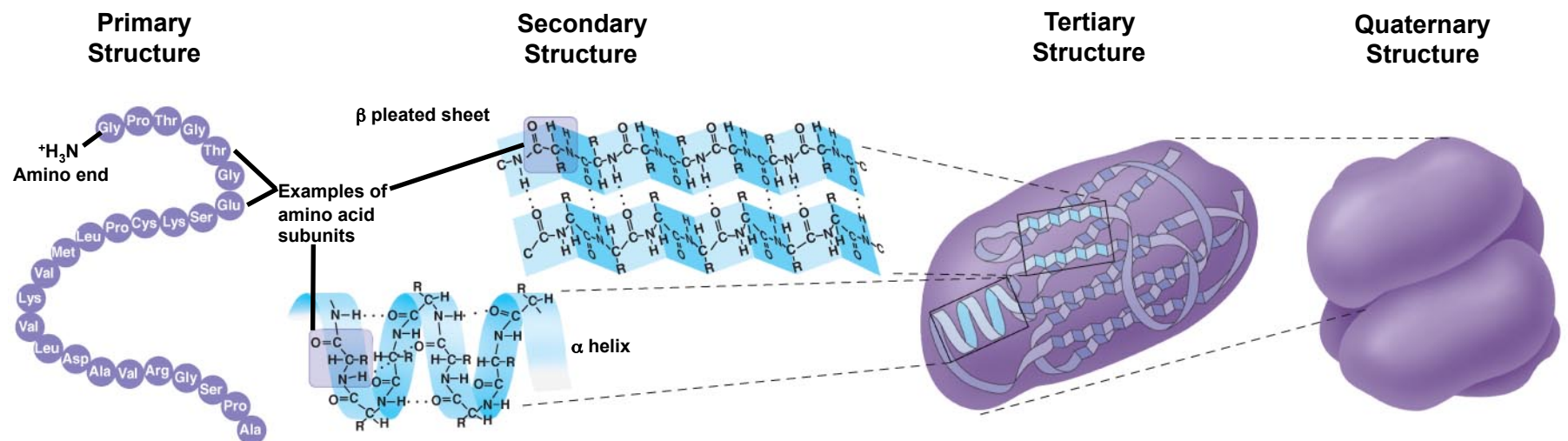
(b) A space-filling model of lysozyme

Antibody protein

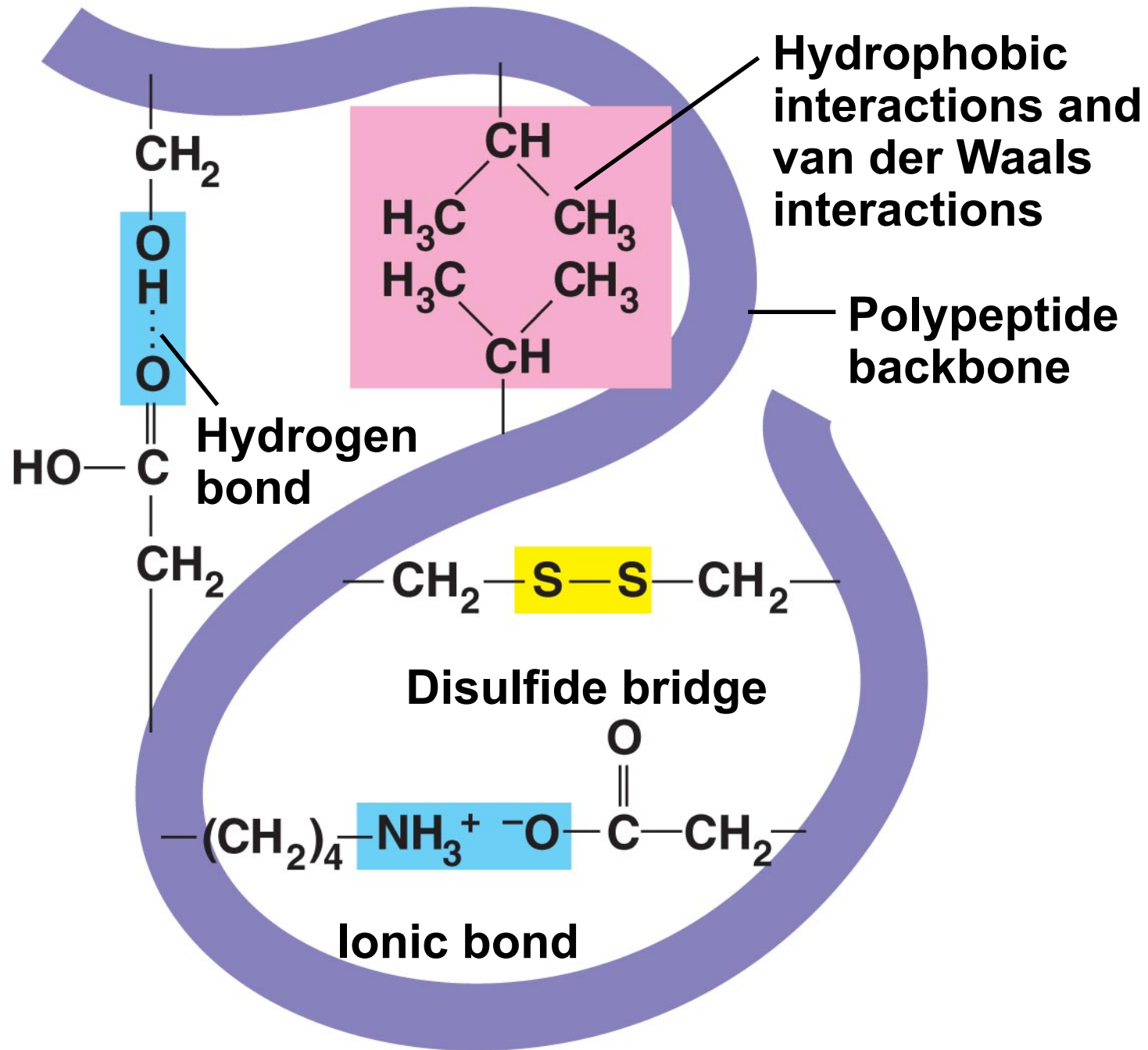
Protein from flu virus

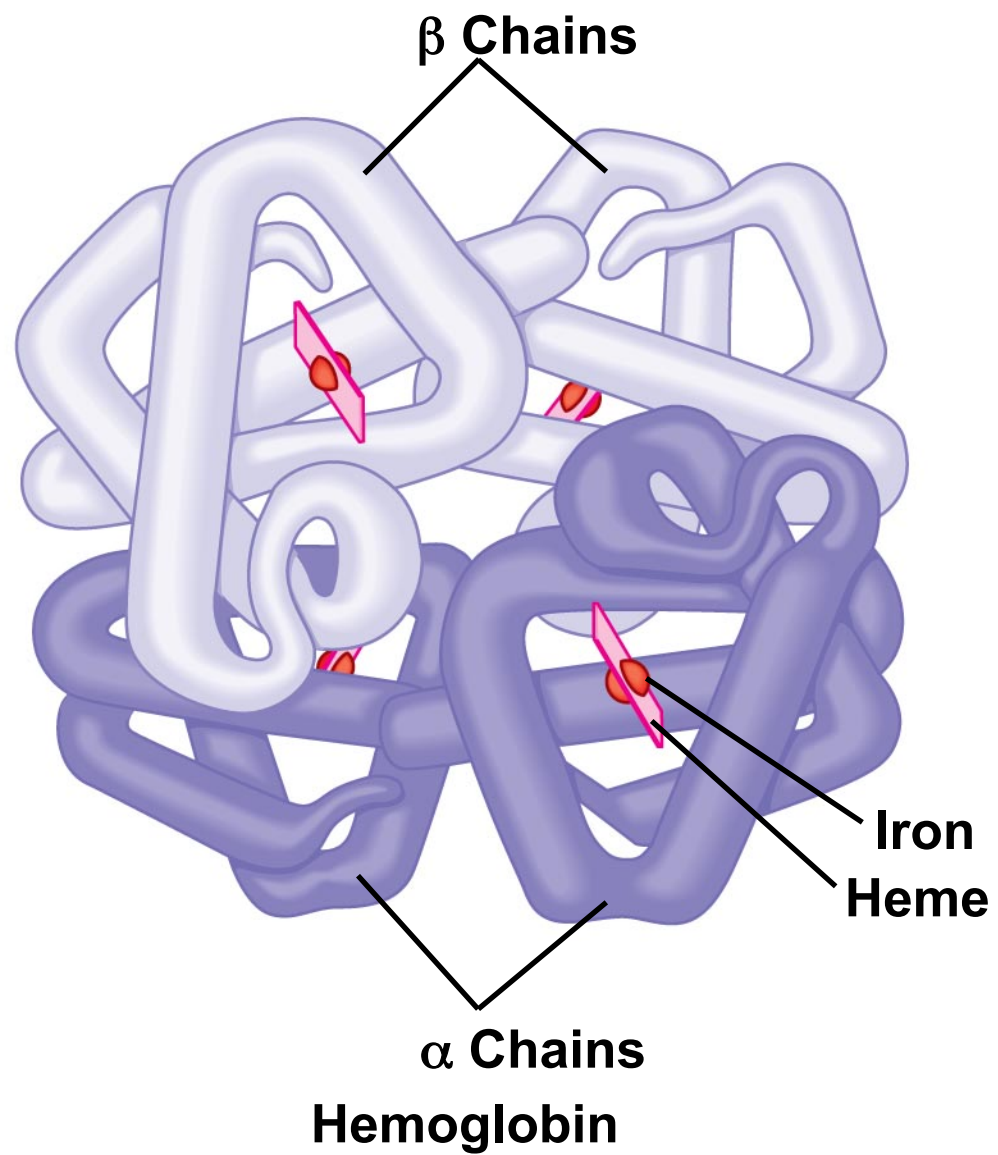
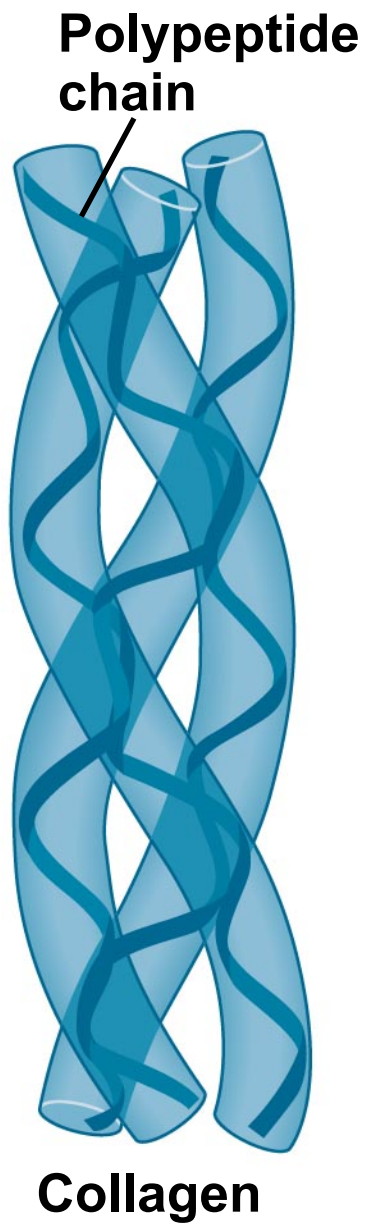


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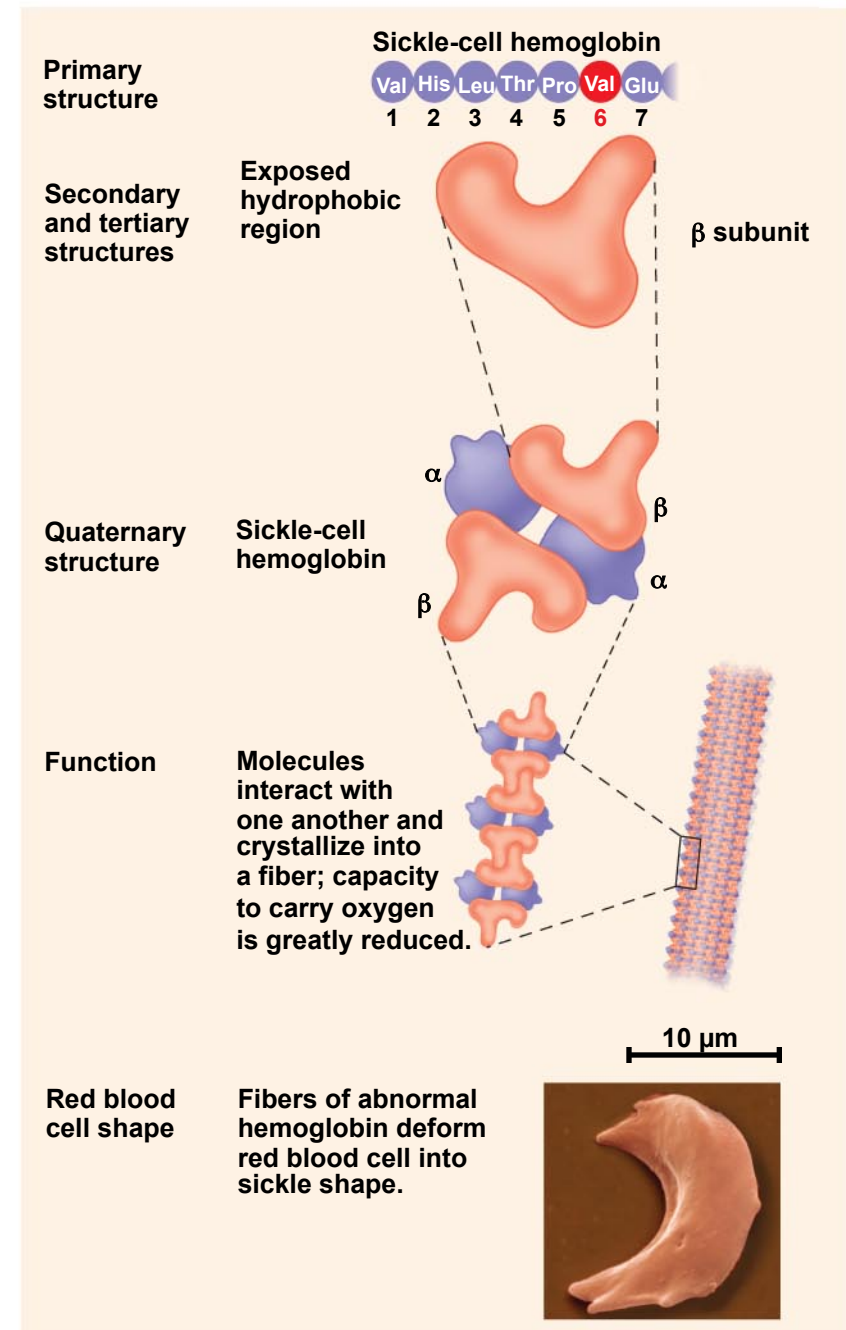
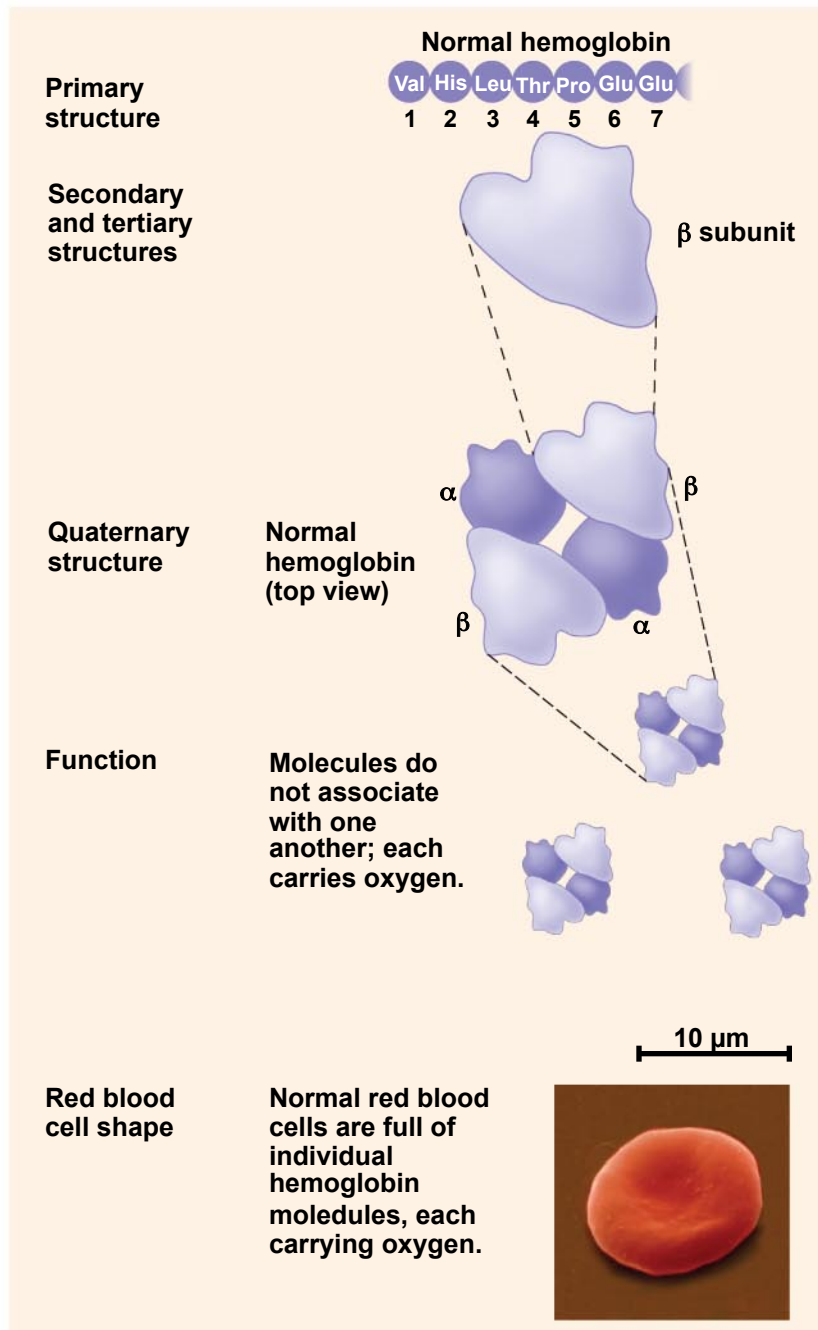


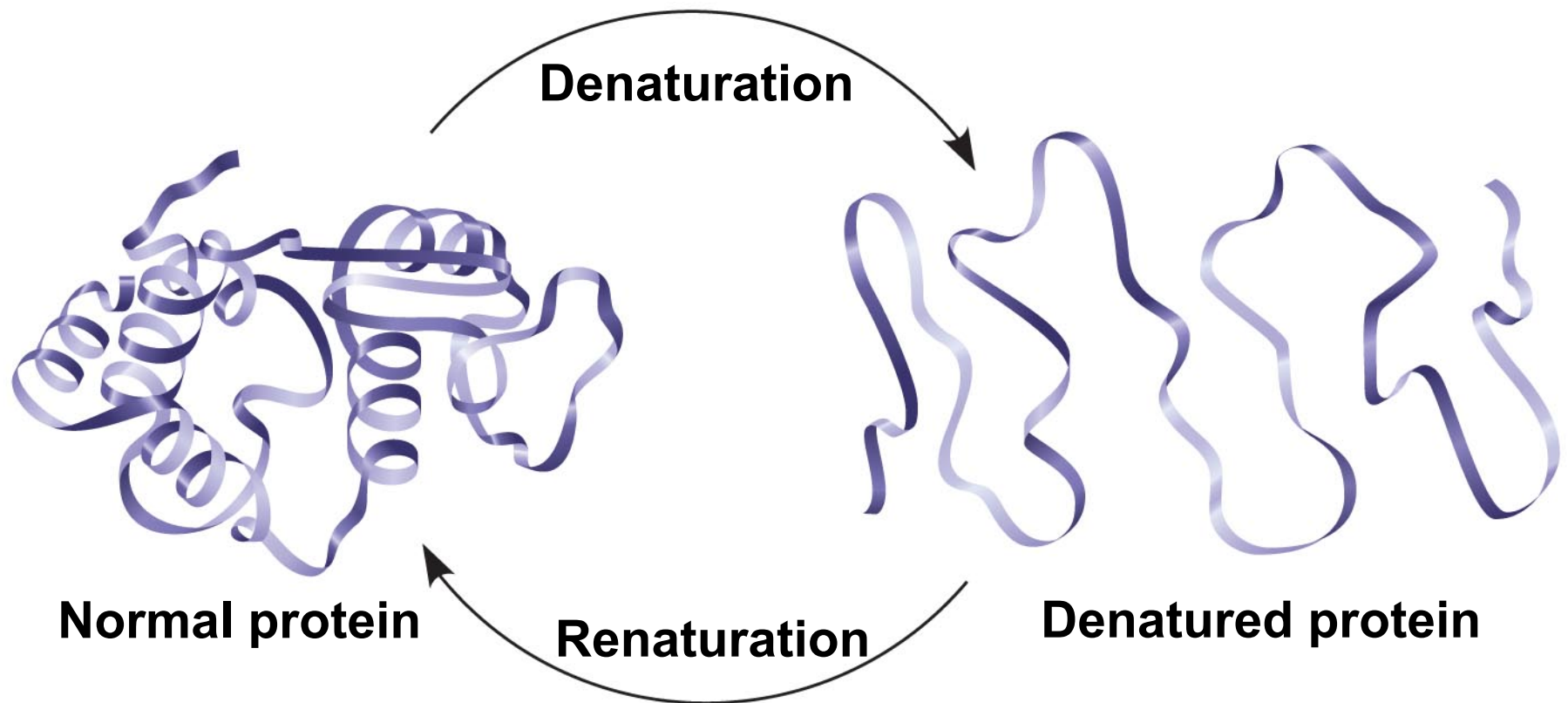
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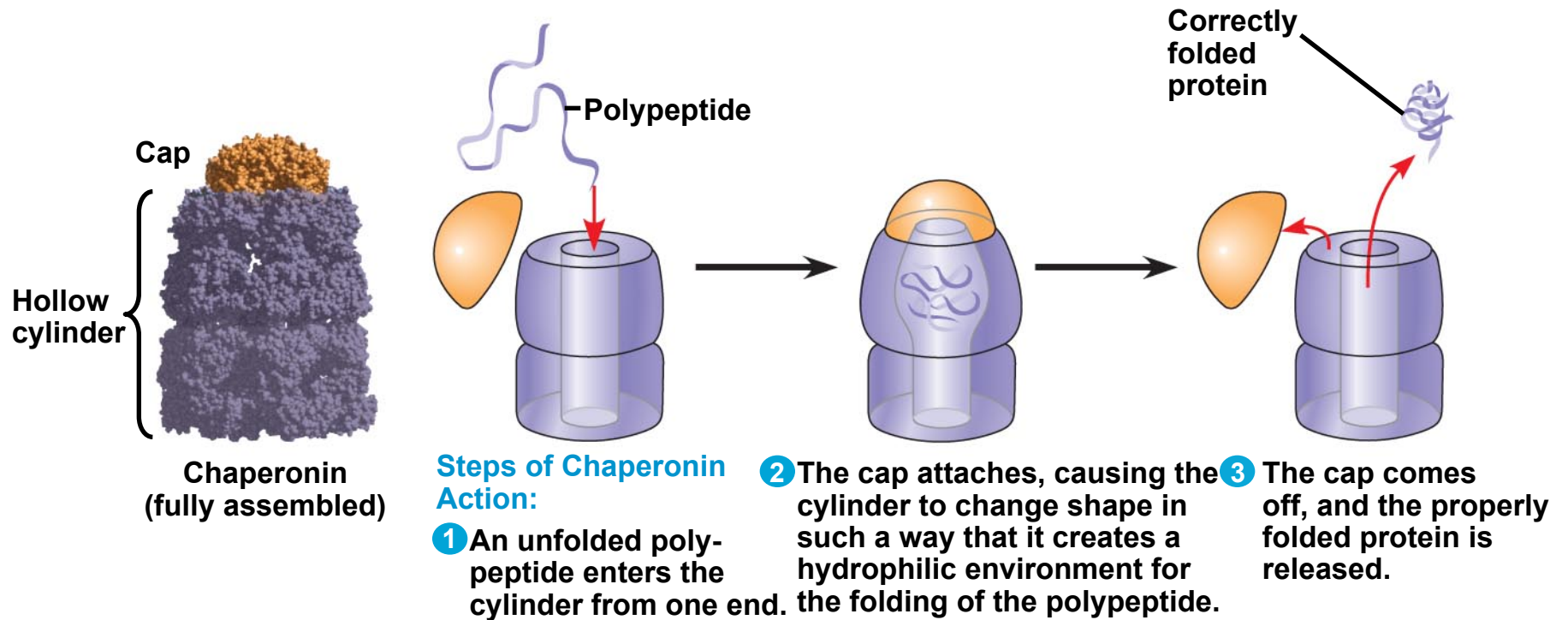


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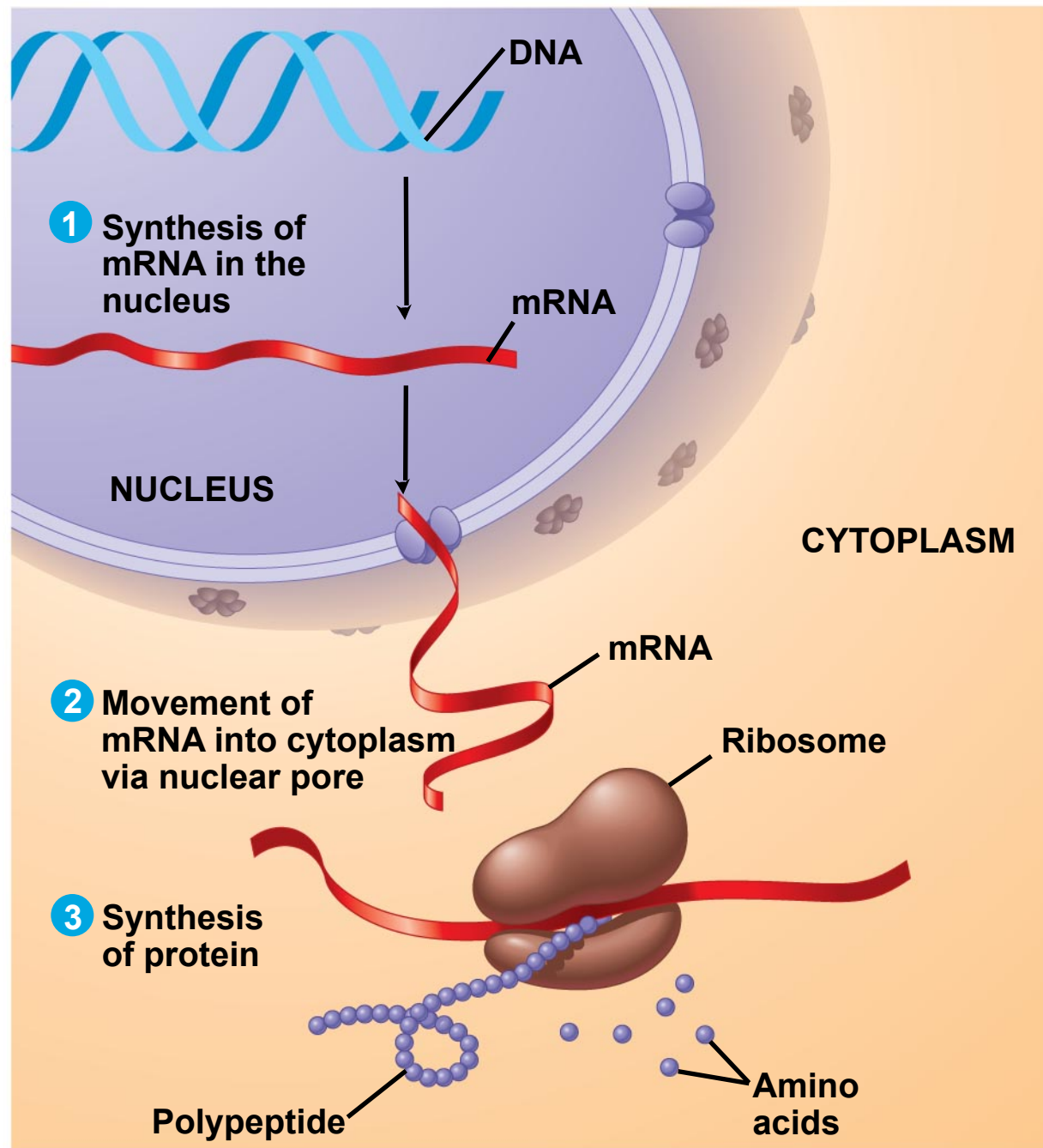




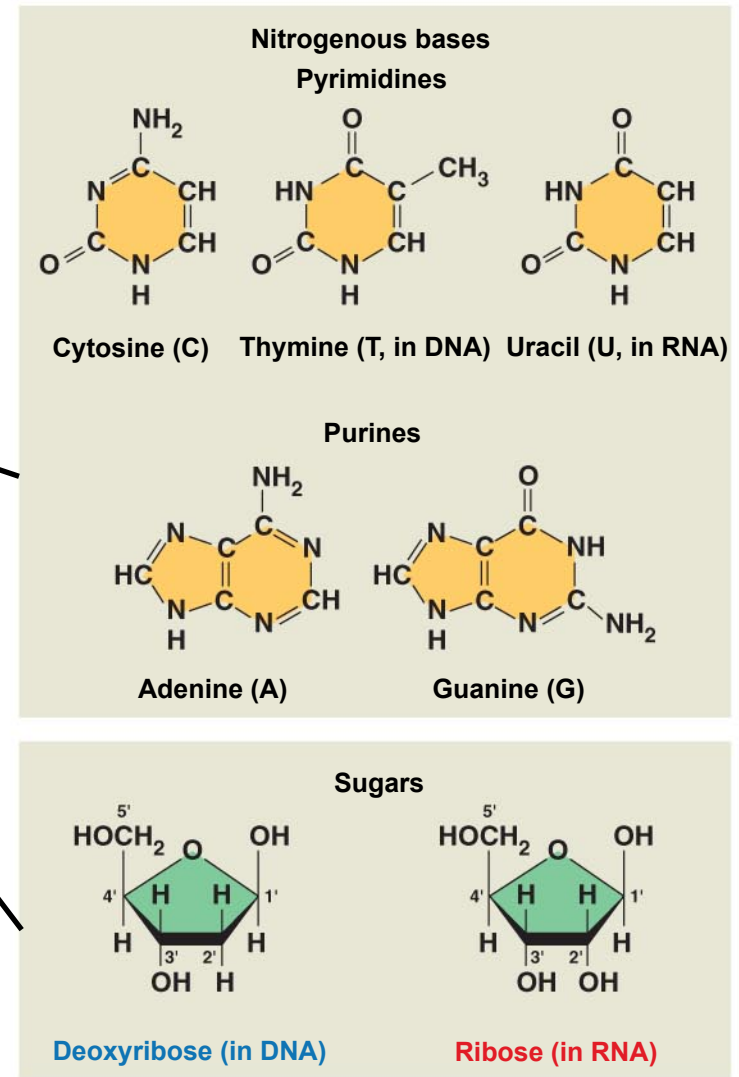
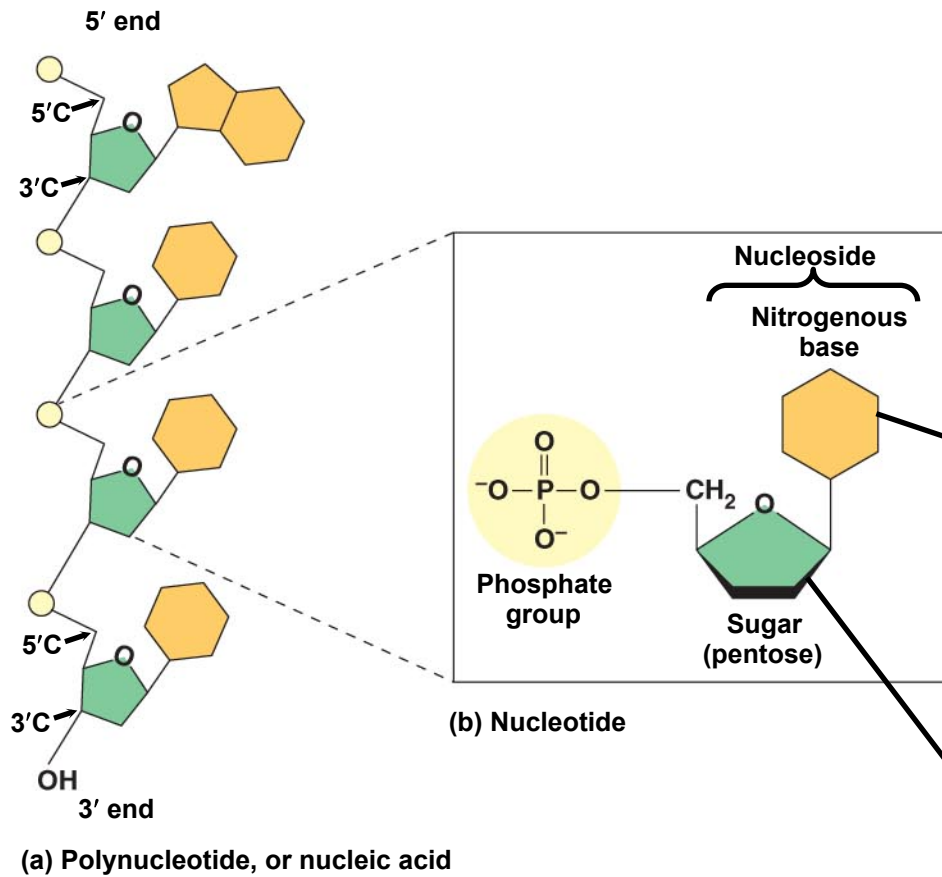
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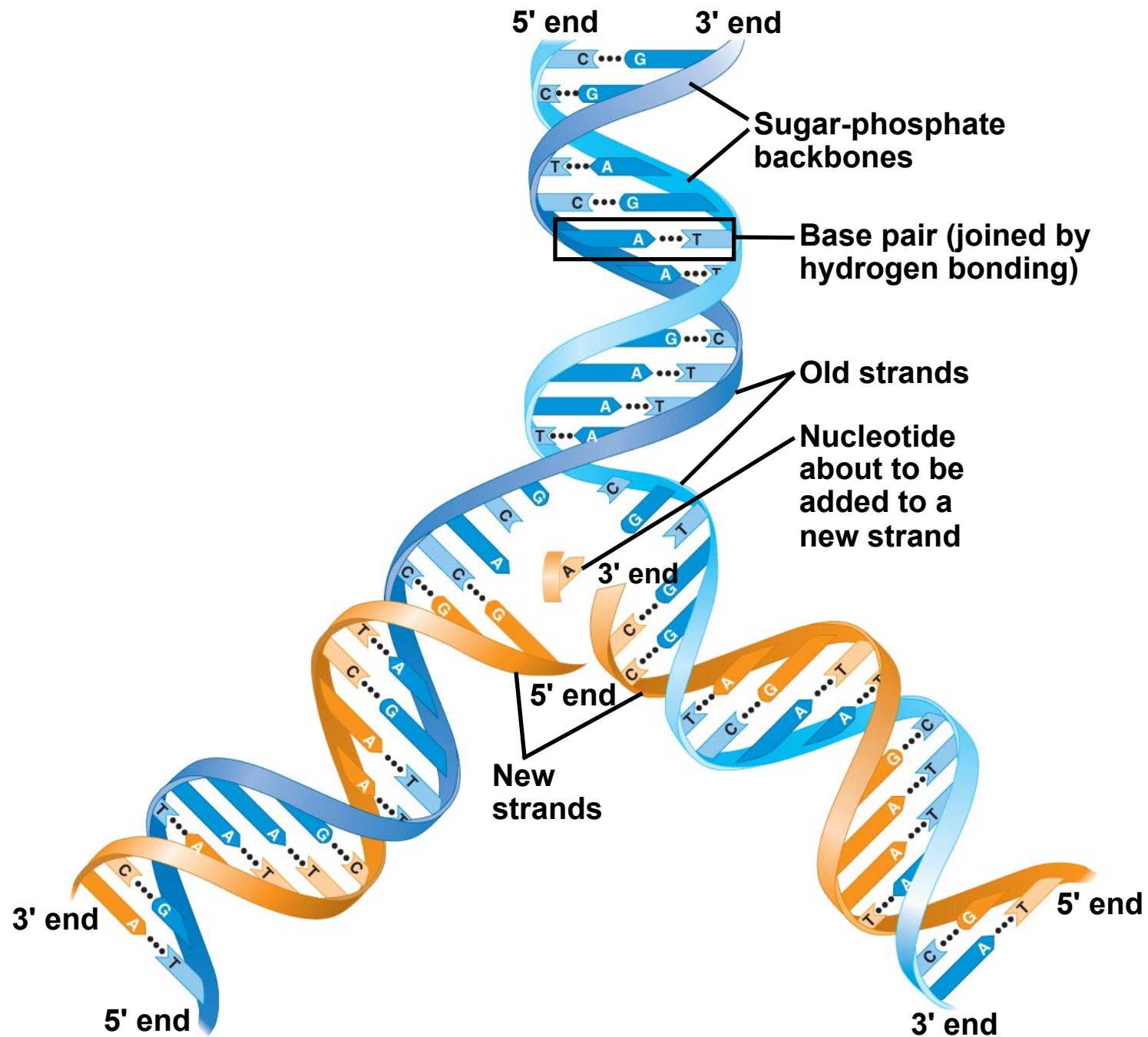
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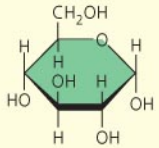


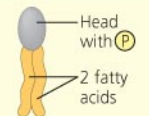
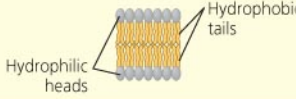

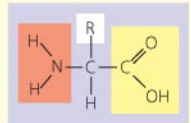
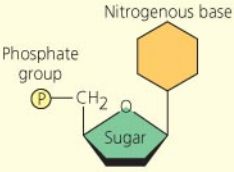




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(c) Nucleoside components: sugars



Large Biological Molecules	Components	Examples	Functions
Concept 5.2 Carbohydrates serve as fuel and building material	 Monosaccharide monomer	Monosaccharides: glucose, fructose	Fuel; carbon sources that can be converted to other molecules or combined into polymers
		Disaccharides: lactose, sucrose	
		Polysaccharides: <ul style="list-style-type: none"> • Cellulose (plants) • Starch (plants) • Glycogen (animals) • Chitin (animals and fungi) 	<ul style="list-style-type: none"> • Strengthens plant cell walls • Stores glucose for energy • Stores glucose for energy • Strengthens exoskeletons and fungal cell walls
Concept 5.3 Lipids are a diverse group of hydrophobic molecules and are not macromolecules	 Glycerol 3 fatty acids	Triacylglycerols (fats or oils): glycerol + 3 fatty acids	Important energy source 
	 Head with (P) 2 fatty acids	Phospholipids: phosphate group + 2 fatty acids	Lipid bilayers of membranes  Hydrophobic tails Hydrophilic heads
	 Steroid backbone	Steroids: four fused rings with attached chemical groups	<ul style="list-style-type: none"> • Component of cell membranes (cholesterol) • Signals that travel through the body (hormones)
Concept 5.4 Proteins have many structures, resulting in a wide range of functions	 Amino acid monomer (20 types)	<ul style="list-style-type: none"> • Enzymes • Structural proteins • Storage proteins • Transport proteins • Hormones • Receptor proteins • Motor proteins • Defensive proteins 	<ul style="list-style-type: none"> • Catalyze chemical reactions • Provide structural support • Store amino acids • Transport substances • Coordinate organismal responses • Receive signals from outside cell • Function in cell movement • Protect against disease
Concept 5.5 Nucleic acids store and transmit hereditary information	 Nitrogenous base Phosphate group Sugar Nucleotide monomer	DNA:  <ul style="list-style-type: none"> • Sugar = deoxyribose • Nitrogenous bases = C, G, A, T • Usually double-stranded 	Stores all hereditary information
		RNA:  <ul style="list-style-type: none"> • Sugar = ribose • Nitrogenous bases = C, G, A, U • Usually single-stranded 	Carries protein-coding instructions from DNA to protein-synthesizing machinery